



	PROTONS	NEUTRONS	ELECTRONS
${}^{63}_{30}\text{Zn}$	30	33	30
${}^{62}_{29}\text{Cu}$	29	33	29
${}^{64}_{28}\text{Ni}$	28	36	28
${}^4_2\text{He}$	2	2	2
${}^{13}_6\text{C}$	6	7	6

	PROTONES	NEUTRONES	ELECTRONES
²³ ₁₁ Na	11	12	11

→ Número Atómico (Z) = 11

11 Electrones

Número Másico = Protones + Neutrones → CASILLA
 (A) (Z) 11

$$23 = 11 + x$$

Protons | Neutrons | Electrons

$^{58}_{28}\text{Ni}$

28

30

28